# Analgesic Narcotic Antagonists. 15.<sup>1</sup> Potent Narcotic Agonist $7\beta$ -(Arylalkyl)-4.5 $\alpha$ -epoxymorphinans

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Structure-activity correlations in  $7\beta$ -(arylalkyl)-3-methoxy- or -hydroxy- $4,5\alpha$ -epoxymorphinans have been investigated.  $6\beta$ -Hydroxy- $7\alpha$ -hydroxymethyl compounds 7 with  $7\beta$ -substituents  $CH_2CH_2R$  [a, R = H; b, R =  $CH_2CH_3$ ; c, R =  $C_6H_5$ ; d, R =  $CH_2C_6H_5$ ; f, R =  $CH_2CH_2C_6H_5$ ; g, R =  $(CH_2)_3C_6H_5$ ; h, R =  $(CH_2)_4C_6H_5$ ] were prepared. Wittig condensations with previously reported  $4,5\alpha$ -epoxy-7 $\beta$ -formyl-7 $\alpha$ -(hydroxymethyl)-6 $\beta$ ,7 $\alpha$ -O-isopropylidene-3methoxy-17-methylmorphinan-6β-ol (3), followed by dilute acid removal of the blocking group and then hydrogenation, gave saturated compounds 7. Compounds with a  $6\alpha$ ,  $7\alpha$ -oxymethylene ring, 18c, d, f, g, were prepared from  $7\beta$ -formyl derivative 16 and the appropriate Wittig reagent, followed by hydrogenation. Both the  $6\beta$ -hydroxy- $7\alpha$ -hydroxymethyl and  $6\alpha$ ,  $7\alpha$ -oxymethylene series containing  $7\beta$ -arylalkyl groups with an alkyl chain length of 2 to 4 are potent narcotic The most potent 17-methyl compound,  $4.5\alpha$ -epoxy- $7\alpha$ -(hydroxymethyl)-17-methyl- $17\beta$ -(4-phenylagonists. butyl)morphinan- $3,6\beta$ -diol (8f) was 700 times more potent than morphine in the acetic acid induced mouse writhing assay. 17-Methyl compounds in the c, d, f, g series were converted to 17-cyclopropylmethyl (P series) or 17cyclobutylmethyl (B series) derivatives. Narcotic antagonistic activity could not be demonstrated for these potent agonist 17-cycloalkylmethyl derivatives. These pharmacological results parallel those previously reported for tertiary alcohol derivatives of the endo-ethenotetrahydrooripavines. Structural considerations confirm the existence of a lipophilic site extending upward and outward from where the C ring of morphine and congeners bind to opiate receptors.

In Part 1 of this series,<sup>2</sup> we described our rational for the incorporation of lipophilic groups into the C ring of the morphine nucleus. These appendages were intended to interact with a lipophilic site hypothesized to exist near the region where the C ring of morphine is bound onto the opiate receptor. The existence of this specific receptor site<sup>3</sup> was advanced as one factor to explain the potent agonist activity observed with tertiary alcohols derived from Diels-Alder adducts<sup>4</sup> of thebaine with methyl vinyl ketone.

In an attempt to utilize this specific receptor site for the preparation of clinically useful opiate analgesic agents, we prepared a series of C8-alkyl derivatives of the morphinan skeleton.<sup>2,5,6</sup> Pharmacological results obtained with these compounds indicated that the lipophilic site was not in the proximity to where C8 is bound to the receptor. Likewise, our studies with C7-alkyl<sup>7,8</sup> and -arylalkyl<sup>8</sup> addends, in which the newly introduced group was in the  $7\alpha$ -position (i.e., equatorial) of the morphinan nucleus, did not lead to potent narcotic agonists. A similar observation was made by Quick et al.<sup>9</sup> for a series of related 7-alkanoylsubstituted hydromorphones. We recently devised methods that allowed the preparation of nonepimerizable  $7\beta$ substituted-dihydromorphine derivatives.<sup>10</sup> To our surprise, the first members of this series in which the substituting 7 $\beta$ -group was phenylbutyl showed potent narcotic agonist activity. The present report describes our efforts directed toward further clarifying structure-activity relationships within this latter series of compounds.

Chemistry. Methods detailing the preparation of the  $7\beta$ -phenylbutyl compounds series **f**, via intermediate e, including studies confirming the structural assignments, have been described by us.<sup>10</sup> We have used this same methodology for the preparation of new compounds, series a-d, g, h, from 1 with the exception that saturated Wittig reagents were used in condensations with aldehydes 3 and 16. Intermediate  $3^{10}$  for the 6 $\beta$ -hydroxy-7 $\alpha$ -hydroxymethyl series (see Scheme I) was prepared from 7,7-bis(hydroxymethyl)dihydroisocodeine  $(1)^{11}$  by reaction with acetone, yielding  $6\beta$ , $7\alpha$ -dioxolane 2 as the major product. Oxidation of the primary alcohol function in 2 with Me<sub>2</sub>SO-TFAA<sup>12</sup> at -60 °C gave the  $7\beta$ -formyl compound 3.

Wittig anions for condensations with 3 were generated from the appropriate triphenylphosphonium salts<sup>13</sup> with





either NaH-Me<sub>2</sub>SO<sup>14</sup> (method A) or phenyllithium<sup>13b</sup> (method B). The initial condensation product 4 was not

- For paper 14, see Kotick, M. P. J. Org. Chem., in press.
   Kotick, M. P.; Leland, D. L.; Polazzi, J. O.; Schut, R. N. J. Med. Chem. 1980, 23, 166.
- (3) Lewis, J. W.; Bentley, K. W.; Cowen, A. Annu. Rev. Pharmacol. 1971, 11, 241.
- (4) For a review of the chemistry of these compounds, see Bentley, K. W. Alkaloids (N.Y.) 1971, 13, 1.
- (5) Polazzi, J. O.; Schut, R. N.; Kotick, M. P.; Howes, J. F.; Osgood, P. F.; Razdan, R. K.; Villarreal, J. E. J. Med. Chem. 1980, 23, 174.
- (6) Kotick, M. P.; Polazzi, J. O. J. Heterocycl. Chem. 1981, 18, 1029.

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 $7\beta$ -(Arylalkyl)-4, $5\alpha$ -epoxymorphinans

### Scheme II



isolated but converted to 5 by removal of the isopropylidene blocking group with dilute acid. Triphenylphosphine oxide was separated from the resultant mixture by selective extraction from the acidic solution. In the preparation of 5d, where a low yield of 4d was obtained by method B, we also isolated  $\alpha$ -hydroxybenzyl derivative 6, as a single isomer. Hydrogenation of 5 to 7, over 10% Pd/C in aqueous EtOH, proceeded slowly. Occasionally, we experienced difficulty in this reduction. The utilization of fresh charges of catalyst and/or chromatography to remove traces of (C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>PO overcame this problem.

The 3-methoxy group was cleaved from 7 to give morphine derivatives 8 by using refluxing 48% HBr. *N*-Cycloalkylmethyl compounds 11 were prepared from *N*methyl derivatives 7 by use of the cyanogen bromide-acid hydrolysis-alkylation sequence previously described.<sup>2</sup>

- (7) Leland, D. L.; Kotick, M. P. J. Med. Chem. 1980, 23, 1427.
  (8) Kotick, M. P.; Leland, D. L.; Polazzi, J. O.; Howes, J. F.; Bousquet, A. R. J. Med. Chem. 1981, 24, 1445.
- (9) Quick, J.; Herlihy, P. Razdan, R. K.; Howes, J. F. J. Med. Chem. 1982, 25, 983. See also Herlihy, P.; Dalzell, H. C.; Howes, J. F.; Razdan, R. F. Ibid. 1982, 25, 986, for some related work.
- (10) Leland, D. L.; Kotick, M. P. J. Org. Chem., in press.
- (11) Leland, D. L.; Kotick, M. P. J. Med. Chem. 1981, 24, 717.
  (12) Omura, K.; Sharma, A. K.; Swern, D. J. Org. Chem. 1976, 41,
- 957. Huang, S. L.; Uman, J. G.; Swern, D. *Ibid.* 1976, 41, 3329. (13) (a) (C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>PCH<sub>3</sub>Br, (C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>P(CH<sub>2</sub>)<sub>2</sub>CH<sub>3</sub>Br, and (C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>PC-
- (H) (Graph, Comparing and Compa
- (14) Corey, E. J.; Chaykovsky, M. J. Am. Chem. Soc. 1965, 87, 1345.
   Greenwald, R.; Chaykovsky, M.; Corey, E. J. J. Org. Chem. 1969, 28, 1128.





Cleavage of the 3-methoxy (11) to the N-cycloalkylmethyl-3-hydroxy compounds (12) was effected with refluxing HBr.

Our general approach to the  $6\alpha$ , $7\alpha$ -oxetane series (see Scheme II) has been described.<sup>10</sup> Briefly, aldehyde 16 was prepared from 1 by the following sequence: tosylation of the primary hydroxy groups to give 13, oxidation of the C6 $\beta$ -hydroxy to a ketone, and, then, borohydride reduction to the  $6\alpha$ -ol 14, which was oxymethylene ring closed in aqueous NaOH-dioxane to 15. Displacement of the remaining tosyloxy group with NaOAc in DMF and then saponification to the alcohol and oxidation gave aldehyde 16. Wittig condensation with 16 gave crude 17c,d,e,g, from which (C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>PO was removed by extraction. Hydrogenation of 17, often with difficulty, yielded 7 $\beta$ -arylalkyl- $6\alpha$ , $7\alpha$ -oxymethylenes 18. Reaction of 16 directly with phenyllithium yielded a mixture of the diastereoisomers 19, which could be resolved by chromatography.

Treatment of 18 with refluxing HBr cleaved both the 3-methoxy function and the oxymethylene ring to give  $7\alpha$ -bromomethyl 3,6 $\alpha$ -diols 20. These intermediates were not further characterized but directly ring closed to the  $6\alpha$ , $7\alpha$ -oxymethylene compounds 21 by base treatment.

N-Cyano compounds 22 were prepared in the usual manner.<sup>2</sup> Due to the acid lability of the oxymethylene ring, hydrolysis of 22 to nor compounds 23 was accomplished under basic conditions. Refluxing 22 with 12 equiv of KOH in aqueous dioxane for 5–7 days gave moderate yields of 23. Further transformations to N-cycloalkylmethyl compounds 24 and 25 were carried out in a manner analogous to that described previously.

Our previous work has shown that morphine compounds having a sp<sup>2</sup> hybridized functionality at C6 often exhibit interesting pharmacological properties.<sup>15</sup> The oxymethylene compounds 24 were, therefore, converted to the 6-oxo analogues by a previously utilized route<sup>10,11</sup> (see Scheme III). Reductive cleavage of the oxymethylene ring in 24 with a 3:1 mixture of LiAlH<sub>4</sub>-AlCl<sub>3</sub> gave  $6\alpha$ hydroxy- $7\alpha$ -methyl- $7\beta$ -arylalkyl compounds 26. Oxidation with Me<sub>2</sub>SO-TFAA gave the C6-oxo compounds 27, which were converted to 3-hydroxy analogues 28 with refluxing HBr.

#### Results

The target N-methyl and N-cycloalkylmethyl compounds prepared in the course of this work were assayed for narcotic agonist activity by using acetic acid induced writhing in the mouse.<sup>16</sup> The results of these assays for N-methyl compounds are recorded in Table I, while those for N-cycloalkylmethyl derivatives are presented in Table II. Selected N-cycloalkylmethyl compounds listed in Table II were also tested for morphine antagonist activity in both the mouse<sup>17</sup> and rat tail-flick<sup>18</sup> systems. None of

(16) Whittle, B. A. Br. J. Pharmacol. 1964, 28, 246.

<sup>(15)</sup> Polazzi, J. O.; Kotick, M. P.; Howes, J. F.; Bousquet, A. R. J. Med. Chem. 1981, 24, 1516.

<sup>(17)</sup> Cowan, A.; Lewis, J. W.; MacFarlane, I. R. Br. J. Pharmacol. 1977, 60, 537.

1052 Journal of Medicinal Chemistry, 1983, Vol. 26, No. 7

Table I. Narcotic Agonist Activity (Mouse Writhing Assay) of N-Methyl Compounds

compd	R	$ED_{50}, \mu mol/kg, sc (95\% CL)^{a}$
5a	H	>28
5d <sup>b</sup>	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	1.1(0.4-2.9)
5g	$(CH_2)_3C_6H_5$	4.0(2.5-6.4)
5h <sup>b</sup>	$(CH_2)_4C_6H_5$	13.9 (11.7-16.4)
6	с	2.3(0.3-17.2)
7a <sup>b</sup>	н	2.7(1.5-4.7)
7b <sup>0</sup>	$CH_2CH_3$	13.7 (7.6-24.8)
7c	C₅H₅	$0.028 (0.009 - 0.085)^d$
7 d <sup>6</sup>	$CH_2C_6H_5$	0.39(0.01-1.4)
7f .	$(CH_{2})_{2}C_{6}H_{5}$	0.014(0.009 - 0.017)
7g <sup>0</sup>	$(CH_2)_3C_6H_5$	2.3(1.6-3.4)
7h	$(CH_{2})_{4}C_{6}H_{5}$	4.5 (1.3-15.9)
8b	$CH_2CH_3$	0.56(0.13-2.1)
8c	C₄H₅	0.004(0.0005 - 0.21)
8d	$CH_2C_6H_5$	0.85 (0.37-1.9)
8f	$(CH_2)_2C_6H_5$	0.003 (0.0024 - 0.0046)
8g 0	$(CH_2)_3C_6H_5$	12.0 (9.6 - 15.0)
8h 0	$(CH_2)_4C_6H_5$	>19
$18c^{e}$	C₅H₅	0.9(0.06-12.1)
18d <i>°</i>	$CH_2C_6H_5$	0.046 (0.03-0.07)
$18f^e$	$(CH_{2})_{2}C_{6}H_{5}$	0.005 (0.001-0.019)
18g <sup>e</sup>	$(CH_2)_3C_6H_5$	1.3(1.1-1.6)
197	с	6.8(2.2-21.1)
197	с	11.9 (5.3-26.9)
$21c^{e}$	C <sub>6</sub> H₅	0.048(0.01-0.17)
21d <sup>e</sup>	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub> 4	0.019(0.008 - 0.047)
$21f^e$	$(CH_2)_2C_6H_5$	0.024(0.015-0.38)
21g <sup>e</sup>	$(CH_2)_3C_6H_5$	0.028 (0.008-0.083)
codeine		10.3(2.7-40)
morphi	ne	2.1(1.1-4.0)
dihydrocodeinone		2.4(1.6-3.6)
dihydro	morphinone	0.25 (0.12-0.44)

<sup>a</sup> Compounds that were prepared as salts (see below) were administered in distilled water; free bases were dissolved by the addition of 1 N HCl and then further diluted. All compounds were administered by the subcutaneous route. <sup>b</sup> HCl salt. <sup>c</sup>  $7\beta$ -( $\alpha$ -Hydroxybenzyl). <sup>d</sup> Repeat 0.014 (0.0023-0.053). <sup>e</sup> d-Tartrate salt. <sup>f</sup> Diastereoisomeric at the substituting  $\alpha$  carbon.

these compounds were, however, active at a dose of 10 mg (ca. 20  $\mu$ mol)/kg in antagonizing the effects of an ED<sub>80</sub> of morphine.

The introduction of unsaturated substituents into the 7 $\beta$ -position of 6 $\beta$ -hydroxy-7 $\alpha$ -(hydroxymethyl)-4,5 $\alpha$ -epoxymorphinans, to give compounds 5, does not enhance narcotic agonist activity. An alkyl group in the same  $7\beta$ position, i.e., 7a and 7b, likewise does not give particularly potent compounds. The arylalkyl compounds 7c-h, in contrast, exhibit agonist activity that is dependent on the length of the methylene bridge between the tertiary C7 carbon and the aryl group. The most potent compounds, 7c and 7f, both contain an even number of carbons in this bridge. Increasing the length of the bridge to  $\geq 5$  in 7g,h results in a decrease in potency. The same general structure-activity trends observed with the 3-methoxy compounds 7 are reflected in the 3-hydroxy series 8. The phenols 8c and 8f are more potent agonists, by about seven- and five-fold, respectively, then their 3-methoxy counterparts. Compound 8f, the most potent N-methyl compound prepared in this study, is approximately 700 times as potent an agonist as morphine in our assay system.

The oxymethylene ring at the  $6\alpha$ , $7\alpha$ -positions enhances agonist potency in 18d,f when compared to 7d,f. The most potent member of this series, 18f, has four methylene units connecting the tertiary carbon and the aryl ring. The

Kotick et al.

Table II. Narcotic Agonist Activity (Mouse Writhing Assay) of N-(Cycloalkylmethyl) Compounds

compd	R	R.	$ED_{so}, \mu mol/kg, sc$ (95% CL) <sup>a</sup>
11D.b			
	C <sub>6</sub> H <sub>5</sub>	$CH_2$ -C-C <sub>3</sub> H <sub>5</sub>	0.61(0.47-0.76)
	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	$CH_2$ -C-C <sub>3</sub> H <sub>5</sub>	0.13(0.05-0.33)
	$(CH_2)_2C_6H_5$	$CH_2$ -c- $C_3H_5$	0.22(0.11-0.43)
11Pg <sup>o</sup>	$(CH_2)_3C_6H_5$	CH <sub>2</sub> -c-C <sub>3</sub> H <sub>5</sub>	0.40(0.25-0.61)
11Bc v	C <sub>6</sub> H₅	$CH_2$ -c- $C_4H_7$	0.70 (0.06-7.3)
TIRd	CH₂C <sub>6</sub> H₅	$CH_2$ -c- $C_4H_7$	0.04(0.007 - 0.15)
11Bf	$(CH_2)_2C_6H_5$	$CH_2$ -c- $C_4H_7$	0.52(0.19-1.4)
11Bg °	$(CH_2)_3C_6H_5$	$CH_2$ -c- $C_4H_7$	0.30 (0.09-1.0)
12Pc	C₅H₅	CH₂-c-C₃H₅	0.95 (0.26-3.5)
12Pd <sup>o</sup>	$CH_2C_6H_5$	CH₂-c-C₃H₅	0.47(0.18 - 1.3)
12Pf	$(CH_2)_2C_6H_5$	$CH_2$ -c- $C_3H_5$	11.0 (6.9-17.6)
12Pg °	$(CH_2)_3C_6H_5$	CH₂-c-C₃H₅	>19
12Bc	C₄H₅	$CH_2$ -c- $C_4H_7$	2.5(1.8 - 3.6)
12Bd_0	CH₂C <sub>6</sub> H₅	$CH_2$ -c- $C_4H_7$	0.57(0.25-1.3)
12Bf <sup>6</sup>	$(CH_2)_2 C_6 H_5$	$CH_2$ -c- $C_4H_7$	21.3(12.4-36.3)
12Bg <sup>o</sup>	$(CH_2)_3C_6H_5$	$CH_2$ -c- $C_4H_7$	>18
24Pc	C <sub>6</sub> H <sub>5</sub>	$CH_2$ -c- $C_3H_5$	0.57(0.22 - 1.5)
24 Pd <sup>c</sup>	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	CH <sub>2</sub> -c-C <sub>3</sub> H <sub>5</sub>	1.07(0.77 - 1.5)
24Pf <sup>c</sup>	$(CH_2)_2C_6H_5$	$CH_2 - c - C_3 H_5$	0.61(0.41-0.93)
24Bc <sup>c</sup>	C <sub>6</sub> H,	CH <sub>2</sub> -c-C <sub>4</sub> H,	>16
24Bdc	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	$CH_2$ -c- $C_4H_2$	4.4(0.69-27.8)
24Bf <sup>c</sup>	$(CH_{2})_{2}C_{6}H_{5}$	$CH_2$ -c- $C_4H_2$	2.8(1.1-7.3)
25Pc	Ċ,H,	$CH_2 - c - C_3H_5$	0.04(0.02-0.07)
25Pd	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	CH <sub>2</sub> -c-C <sub>3</sub> H	0.003(0.001-0.1)
25 Pf <sup>c</sup>	$(CH_{2})_{2}C_{6}H_{5}$	$CH_2$ -c- $C_3H_3$	0.015 (0.006-0.3)
25Bc	C <sub>6</sub> H,	$CH_2$ -c- $C_4H_2$	0.39(0.25-0.55)
25Bd <sup>c</sup>	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	$CH_2 - c - C_4H_7$	0.06(0.048-0.08)
25Bf	$(CH_{2}), C, H_{3}$	$CH_2$ -c- $C_4H_2$	0.05 (0.018-0.10)
26Pf <sup>b</sup>	$(CH_{2}), C_{6}H_{3}$	CH <sub>2</sub> -c-C <sub>3</sub> H <sub>5</sub>	0.50 (0.15-1.5)
26Bc <sup>b</sup>	Ċ,H,	CH, -c-C <sub>4</sub> H,	1.0(0.78-1.3)
26Bf <sup>b</sup>	$(CH_{1})_{1}C_{2}H_{2}$	CH,-c-C,H,	11.3 (8.2-15.6)
27 Pf <sup>b</sup>	(CH_),C_H	CH,-c-C,H.	2.0(1.1-3.5)
27Bc	Ċ.H.	CHc-C_H_	0.19(0.008-0.05)
27Bf	(CH,),C,H,	CH -c-C H.	5.6(1.8-16.9)
28 <b>P</b> f	(CH.),C.H.	CHc-C.H.	0.11(0.10-0.19)
28Bc <sup>b</sup>	C.H.	CHc-C.H.	0.04(0.02-0.11)
28Bf	(ČH <sub>2</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	$CH_2$ -c- $C_4H_7$	0.17 (0.06-0.39)

<sup>a</sup> Compounds that were prepared as salts (see below) were administered in distilled water; free bases were dissolved by the addition of 1 N HCl and then further diluted. All compounds were administered by the subcutaneous route. <sup>b</sup> HCl salt. <sup>c</sup> d-Tartrate salt.

corresponding 3-phenols 21 are all potent agonists with insufficient differences in  $ED_{50}$ 's to allow conclusions to be drawn. The presence of an additional alcohol function at the  $\alpha$  carbon on the C7 $\beta$  side chain in 6 or 19 does not make a contribution to agonist potency.

The more potent N-methyl- $7\beta$ -arylalkyl compounds that were obtained in the initial phases of this work were converted to the corresponding N-cyclopropylmethyl (P series) or N-cyclobutylmethyl (B series) analogues. In both series 11 and 12, the homologue with a three methylene-C7-aryl bridge are the most potent members of these cycloalkylmethyl series. For oxymethylene compounds 24P, the most potent members are 24Pc and 24Pf; in the 24B series, none of the compounds are particularly potent. Potency in the 3-hydroxy series, 25P and 25B, likewise peaks when the C7 and aryl groups are separated by three methylenes. The most potent member of this series, 25Pd, is ca. 700 times as potent an agonist as is morphine.

Compounds with sp<sup>2</sup> hybridization at C6, **27** and **28**, were potent agonists. A comparison of these *N*-cycloalkylmethyl series with the corresponding *N*-cycloalkylmethyl- $6\beta$ -hydroxy- $7\alpha$ -hydroxymethyl series 11 and 12, or oxymethylene compounds **24** and **25**, reveals that intricate structure-potency relationships exist within these compounds.

To summarize, both the  $6\beta$ -hydroxy- $7\alpha$ -hydroxymethyl and  $6\alpha$ , $7\alpha$ -oxymethylene series containing  $7\beta$ -arylalkyl

<sup>(18)</sup> Harris, L. S.; Pierson, A. K. J. Pharmacol. Exp. Ther. 1964, 143, 141.

 $7\beta$ -(Arylalkyl)-4, $5\alpha$ -epoxymorphinans



groups with an alkyl chain length of two to four methylene units are potent agonists. Narcotic antagonist activity was not demonstrated for any compounds prepared in the course of the present work.

## Discussion

The previous discovery of potent narcotic agonist activity in the homologous series of C19 tertiary alcohols 29 (Chart I) resulted in the hypothesis<sup>3</sup> that a specific site exists on the opiate receptor for the lipophilic groups that had been incorporated into this region of the morphine molecule. Structure-activity studies revealed that agonist potency for 29 is maximal when n = 2 or 3. Further increases in n result in a dimunition of potency. This potency profile is even more pronounced in the analogous series 30 in which a phenyl group has been placed at the end of the alkyl chain. In series 30, there is over a 1000fold increase in potency when the methylene bridge is extended from n = 0 to n = 1, a further 10-fold increase between n = 1 and n = 2, and, then, a 10-fold decrease in potency between n = 2 and n = 3. In the related series 31,<sup>19</sup> which most closely resembles our presently reported compounds in structure, potency is likewise maximized when n = 2 or 3.

Prior consideration of the potencies of compounds 30, which contain terminal aryl groups, allowed the lipophilic site to be mapped onto the putative opiate receptor surface. The maximal agonist potency observed in 30 when n = 2 places this binding site at a distance centered about 6 Å from C7 of the receptor-bound morphine molecule.<sup>20</sup> The pharmacological results obtained during the present work are in agreement with these previous findings. A comparison of the molecular models of 32 and 33, in which n = 3 or 4 and for which the methylene units are in a fully extended configuration, with 29-31 places the aryl group within the same locale on a receptor surface.

The presence of an alcohol function at C19 is not a prerequisite for potent agonist activity in the *endo*ethenotetrahydrooripavines.<sup>20</sup> Hydrogen bonding between the 6-methoxy group and the tertiary C19 alcohol has likewise been eliminated as contributing to the potencies of **29** and **30**.<sup>21</sup> This finding is further confirmed by this work. The oxetane ring in **33** does not appreciably alter

- (19) Lewis, J. W.; Readhead, M. J. J. Med. Chem. 1973, 16, 84.
- (20) Bentley, K. W.; Lewis, J. W. in "Agonist and Antagonist Actions of Narcotic Drugs"; Kosterlitz, H. W.; Collier, H. O. J.; Villarreal, J. E., Eds.; University Park Press: Baltimore, MD, 1973, p 7.
- (21) Hutchins, C. W.; Cooper, G. K.; Pürro, S.; Rapoport, H. J. Med. Chem. 1981, 24, 773.

potency; likewise, our potent  $7\alpha$ -methyl compounds 27 and 28 cannot form hydrogen bonds due to the absence of an exocyclic hydroxy function.

The most significant structural difference between 29–31 and 32–33 is the absence of the 6,14-etheno bridge. The C ring in 29–31 is fixed by  $C_{17}$ – $C_{18}$  into a boat conformation. This etheno bridge results in a rigid, cage-like structure in this region of the morphine molecule. The presently reported compounds are not so constrained. The oxetane ring in 33 does limit the flexibility of the C ring. Series 32 are, however, not so constrained but may have a conformational preference in solution imposed by the presence of the various C7-substituents.

These observations, together with our previous work<sup>1,22</sup> and those with 2,6-methano-3-benzazocine-11-propanols<sup>23</sup> which lack a rigid C-ring structure, diminishes the contribution of ring C rigidity to the agonist potencies observed with **29–31**. This allows the conclusion that the main contributory factor to the potencies of **29–33** and related compounds is the presence of an exocyclic lipophilic aromatic group in the morphine C-ring region extending up and outward from C7. This aromatic binding site has also been implicated in the explanation of the pharmacological activities observed with enkephalins<sup>24</sup> and other structural classes of narcotic analgesic agents.<sup>25</sup>

The lack of narcotic antagonist activity in N-cycloalkylmethyl derivatives of **32** and **33**, in retrospect, is not surprising. Morphine antagonist activity in **29** is restricted to N-cycloalkylmethyl derivatives in which the groups on C19 are small.<sup>3,20</sup> Attempts are being made to explain this phenomenon.<sup>26</sup>

This paper completes our work in the area of analgesic narcotic antagonists. The at first elusive lipophilic site on the opiate receptor, which we initially postulated as existing near C8, has been found extending outward from the  $7\beta$ -position (7-up) of the morphine nucleus. This is exactly where model-building studies with **29–31** indicated it could be found.

### **Experimental Section**

Methods have been previously described.<sup>2</sup> Processing in the usual manner implies that the combined organic phases were washed with dilute NH<sub>4</sub>OH, dried (MgSO<sub>4</sub>), and filtered, and the filtrate was evaporated at a 40-45 °C bath temperature. The residue was further dried at 50-60 °C bath temperature under high vacuum. Column chromatography was carried out over silica gel 60 G (E. Merck) with CHCl<sub>3</sub>-MeOH mixtures containing 0.25 to 1%, v/v, concentrated NH<sub>4</sub>OH. All compounds gave NMR and mass spectra consistent with the indicated structures and in accord with those which we previously reported.<sup>10</sup>

 $7\beta$ -(Arylalkylidene)-4,5 $\alpha$ -epoxy-7 $\alpha$ -(hydroxymethyl)-3methoxy-17-methylmorphinan-6 $\beta$ -ols (5). Method A. A suspension of 50% NaH (0.66 g, 13.8 mmol) in mineral oil was washed three times with hexane while under an argon atmosphere. Me<sub>2</sub>SO (10 mL) was added, and the mixture heated at 60–70 °C until the evolution of H<sub>2</sub> ceased (ca. 30 min). The mixture was cooled to 25 °C, and the appropriate phosphonium salt (13.8 mmol) in Me<sub>2</sub>SO (50 mL) was added dropwise. After 10 min, 3 (5.0 g, 12.5 mmol) in Me<sub>2</sub>SO (50 mL) was added rapidly dropwise. Stirring was continued for 30 min at room temperature, followed by heating of the mixture at 65–70 °C for 30 min. The cooled mixture was diluted with water and extracted with toluene. The

- (23) For the leading reference in this area, see Michne, W. F.; Lewis, T. R.; Michalec, S. J.; Pierson, A. K.; Rosenberg, F. J. J. Med. Chem. 1979, 22, 1158.
- (24) Morley, J. S. Annu. Rev. Pharmacol. 1980, 20, 81.
- (25) For a discussion of this concept, see Portoghese, P. S.; Alreja, B. O.; Larson, D. L. J. Med. Chem. 1981, 24, 782.
- (26) Loew, G. H.; Berkowitz, D. S. J. Med. Chem. 1979, 22, 603.

<sup>(22)</sup> Kotick, M. P. J. Med. Chem. 1981, 24, 722.

Table III.	66-Hv	droxv	-7α-(h	vdrox	vmethvl	l)-7	B-substituted-4	.5α-ep	oxymor	ohinans
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compd	R	R <sub>1</sub>	yield, %	mp, °C	recrystn solvent <sup>a</sup>	formula <sup>b</sup>
7a	Н		58 c	200-204	E	C <sub>21</sub> H <sub>29</sub> NO <sub>4</sub> ·HCl
7b	$CH_2CH_3$		53°	>270	Е	C <sub>23</sub> H <sub>33</sub> NO₄·HCl
7 c	C <sub>6</sub> H,		$72^{d}$	240-242	E	C <sub>27</sub> H <sub>33</sub> NO <sub>4</sub>
7d	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>		81 <i>°</i>	267-270	M-EA	C <sub>28</sub> H <sub>35</sub> NO <sub>4</sub> ·HCl
7f	$(CH_2)_2C_6H_5$		89 <i>°</i>	196-198	EA-C	$C_{29}H_{37}NO_{4}$
7g	$(CH_2)_3C_6H_5$		83°	200-202	M-EA	C <sub>30</sub> H <sub>39</sub> NO <sub>4</sub> ·HCl
7h	$(CH_2)_4C_6H_5$		$67^{d}$	160-161	$\mathbf{E}$	$C_{31}H_{41}NO_{4}$
8b	$CH_2CH_3$		81 <sup>d</sup>	224-226	M-EA	$C_{22}H_{31}NO_{4}$
8c	C₄H₅		54 <sup>a</sup>	259-262	M-EA	$C_{26}H_{31}NO_4$
8d	$CH_2C_6H_5$		$62^{a}$	240-245	$\mathbf{E}$	$C_{27}H_{33}NO_{4}$
<b>8f</b>	$(CH_2)_2C_6H_5$		82 <sup>d</sup>	192-193	$\mathbf{E}\mathbf{A}$	$C_{28}H_{35}NO_{4}$
8g	$(CH_2)_3C_6H_5$		88 <sup>a</sup>	212 - 216	M-EA	C₂9H37NO₄·HCl
<b>8</b> h	$(CH_2)_4C_6H_5$		67 <sup>a</sup>	249-251	$\mathbf{E}$	C <sub>30</sub> H <sub>39</sub> NO <sub>4</sub> ·HCl·H <sub>2</sub> O
11Pc	C₅H₅	CH <sub>2</sub> -c-C <sub>3</sub> H <sub>5</sub>	85 <sup>d</sup>	261-264	E	C <sub>30</sub> H <sub>37</sub> NO₄·HCl
11 <b>P</b> d	$CH_2C_6H_5$	CH₂-c-C₃H₅	$44^{T}$	198-199.5	E	$C_{31}H_{39}NO_4$
11 <b>P</b> f	$(CH_{2})_{2}C_{6}H_{5}$	$CH_2$ -c- $C_3H_5$	63 <sup>a</sup>	256-258	M-EA	C <sub>32</sub> H <sub>41</sub> NO₄·HCl
11Pg	$(CH_{2})_{3}C_{6}H_{5}$	$CH_2$ -c- $C_3H_5$	$75^{a}$	240-243	E	C₃₃H₄₃NO₄·HCl
11Bc	C₄H₅	$CH_2$ -c- $C_4H_7$	$74^{a}$	~200 <sup>g</sup>	$\mathbf{E}$	C <sub>31</sub> H <sub>39</sub> NO₄·HCl
11 <b>B</b> d	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	$CH_2$ -c- $C_4H_7$	$62^{a}$	152 - 154	$\mathbf{E}\mathbf{A}$	$C_{32}H_{41}NO_4$
11Bf	$(CH_2)_2C_6H_5$	$CH_2$ -c- $C_4H_7$	$62^{a}$	256-258	M-EA	C₃₃H₄₃NO₄·HCl
11Bg	$(CH_2)_3C_6H_5$	$CH_2$ -c- $C_4H_7$	$77^{a}$	247-250	$\mathbf{E}$	C <sub>34</sub> H <sub>45</sub> NO <sub>4</sub> ·HCl
12Pc	C₅H₅	$CH_2$ -c- $C_3H_5$	$34^{a}$	252 - 254	E	$C_{29}H_{35}NO_{4}$
12Pd	$CH_2C_6H_5$	$CH_2$ -c- $C_3H_5$	36 <sup>a</sup>	>200  dec	M-EA	C <sub>30</sub> H <sub>37</sub> NO₄·HCl
12Pf	$(CH_2)_2C_6H_5$	$CH_2$ -c- $C_3H_5$	86 <sup>a</sup>	214 - 216	E-A	$C_{31}H_{39}NO_4$
12Pg	$(CH_2)_3C_6H_5$	$CH_2$ -c- $C_3H_5$	$57^{a}$	168 - 174	M-EA	$C_{32}H_{41}NO_4 \cdot HCl^n$
12Bc	C <sub>6</sub> H <sub>5</sub>	$CH_2$ -c- $C_4H_7$	77	255-257	E	C <sub>30</sub> H <sub>37</sub> NO <sub>4</sub>
12Bd	$CH_2C_6H_5$	$CH_2$ -c- $C_4H_7$	49	>280	M-EA	C <sub>31</sub> H <sub>39</sub> NO₄·HCl
12Bf	$(CH_2)_2C_6H_5$	$CH_2$ -c- $C_4H_7$	95	222-225	M-EA	C <sub>32</sub> H <sub>41</sub> NO <sub>4</sub> ·HCl
12Bg	$(CH_2)_3C_6H_5$	$CH_2$ -c- $C_4H_7$	93	182-186	M-EA	C <sub>33</sub> H <sub>13</sub> NO <sub>4</sub> ·HCl

 ${}^{a}$  C = chloroform; E = ethanol; EA = ethyl acetate; M = methanol; W = water.  ${}^{b}$  Compounds had C, H, and N analyses within ±0.4% of the calculated values.  ${}^{c}$  HCl salt crystallized directly.  ${}^{d}$  Yield of free base after chromatography.  ${}^{e}$  Free base crystallized directly.  ${}^{f}$  Directly crystallized from toluene.  ${}^{g}$  Softens.  ${}^{h}$  C: calcd, 71.16; found, 70.69.

organic phase was evaporated, and the residue, consisting of 4 and  $(C_6H_5)_3PO$ , dissolved in EtOH (200 mL) and 1 N HCl (50 mL) was added. The mixture was gently boiled on the steam bath for 30 min and then evaporated to a small volume. The acidic concentrate was diluted with H<sub>2</sub>O and washed several times with toluene. The aqueous solution was then made basic with concentrated NH<sub>4</sub>OH, and further processing was carried out as described below.

Method B. To a suspension of the phosphonium salt (23 mmol) in Et<sub>2</sub>O (200 mL), under argon at room temperature, was added phenyllithium (23 mmol, 1.9 M solution in 7:3 C<sub>6</sub>H<sub>6</sub>-Et<sub>2</sub>O), and the mixture was stirred for 1-2 h. A solution of 3 (4.0 g, 10 mmol) in 1:1 toluene-Et<sub>2</sub>O (100 mL) was added to the dark solution, and stirring was continued for 2 h. The reaction was quenched by the addition of H<sub>2</sub>O, concentrated NH<sub>4</sub>OH was added, and the intermediate 4 was extracted with CHCl<sub>3</sub>. Evaporation of the organic phase was followed by the addition of EtOH and 1 N HCl, and the mixture was boiled for 30 min and then further processed as described below.

Compound 5a was prepared by method A. Crystals of 5a precipitated from the basic aqueous solution and were obtained in 78% yield. Recrystallization from  $H_2O$  gave pure 5a, mp 181–182 °C. Anal. ( $C_{21}H_{27}NO_4$ ) C, H, N. Intermediate 5b was prepared by method B and not further purified but directly hydrogenated to 7b. In a similar manner, 5c was prepared by method A and purified by chromatography to give an 89% yield of the desired product as a foam. Compound 5d was prepared by method B and purified by chromatography to give a 38% yield of 5d as a foam. The HCl salt, mp >280 °C, was obtained in crystalline form from EtOH. Anal. (C28H33NO4 HCl) C, H, N. The preparation of 5e and 5f has been reported.<sup>10</sup> Compound 5g was prepared by method B and obtained as a foam, after chromatography, in 68% yield. Crystallization from EtOH gave pure 5g, mp 168-170 °C. Anal. (C30H37NO4) C, H, N. Compound 5h was prepared by method A and purified by chromatography to give a 70% yield of 5h as a foam. The HCl salt, mp >260 ° was obtained in crystalline form from EtOH. Anal. (C<sub>31</sub>H<sub>39</sub>N-O₄·HCl) C, H, N.

 $4,5\alpha$ -Epoxy-7 $\beta$ - $(\alpha$ -hydroxyben zyl)-7 $\alpha$ -(hydroxymethyl)-3methoxy-17-methylmorphinan-6 $\beta$ -ol (6). This compound was obtained as a side product in the preparation of 5 by method B. Material obtained by chromatography from several reactions, which migrated as a single spot on TLC, was combined and crystallized from MeOH-EtOAc to give a sample of pure 6, mp 265-267 °C. Anal. ( $C_{26}H_{31}NO_5$ ) C, H, N.

 $7\beta$ -(Arylalkyl)-4, $5\alpha$ -epoxy- $7\alpha$ -(hydroxymethyl)-3-methoxy-17-methylmorphinan-6 $\beta$ -ols (7). Hydrogenation of 5, as the free base or HCl salt, was carried out over 10% Pd/C (10-25%, w/w) at 50 psi in aqueous EtOH acidified with HCl (ca. pH 2) until the uptake of H<sub>2</sub> ceased (2-24 h). After the catalyst was removed by filtration, the filtrate was evaporated to a crystalline residue. In cases where crystallization did not occur, the HCl salt was converted to the free base, and further purification was carried out by chromatography or crystallization as indicated in Table III.

 $7\beta$ -(Arylalkyl)-4,5 $\alpha$ -epoxy-3-hydroxy-7 $\alpha$ -(hydroxymethyl)-17-methylmorphinan-3,6 $\beta$ -diols (8). A mixture of 7 (free base or HCl salt) and concentrated HBr (1.0 g in 10–15 mL) was immersed in a preheated oil bath (ca. 140 °C) and refluxed for 10 to 20 min. The reaction mixture was cooled, diluted with H<sub>2</sub>O, and adjusted to pH 10–11 by the addition of concentrated NH<sub>4</sub>OH. The basic solution was extracted with three portions of CHCl<sub>3</sub>, the organic extracts were processed in the usual manner, and the residue was chromatographed. Crystals of the free base or HCl salt were obtained from the solvents indicated in Table III.

 $7\beta$ -(Arylalkyl)-17-cyano-4, $5\alpha$ -epoxy- $7\alpha$ -(hydroxymethyl)-3-methoxymorphinan- $6\beta$ -ols (9). To a rapidly stirred mixture of 7 (1.0 equiv) in CHCl<sub>3</sub> (1 g in 15 mL) containing K<sub>2</sub>CO<sub>3</sub> (1.5 equiv) was added dropwise a solution of BrCN (1.2 equiv) in CHCl<sub>3</sub> (1 g in 15 mL). The mixture was stirred at room temperature for 30 min and then refluxed for 2 h. The insoluble material was removed by filtration, and the filtrate was evaporated. The residue was evaporated with EtOH until a foam formed. This foam, obtained in nearly quantitative yield and homogeneous by TLC, was hydrolyzed to 10 as described below.

 $7\beta$ -(Arylalkyl)-4,5 $\alpha$ -epoxy- $7\alpha$ -(hydroxymethyl)-3-methoxymorphinan-6 $\beta$ -ols (10). A mixture of 9 and 2 N HCl (1 g in 15-25 mL) was refluxed for 8 to 18 h. The solution was cooled, made basic by the addition of concentrated NH<sub>4</sub>OH, and extracted with CHCl<sub>3</sub>. Processing of the CHCl<sub>3</sub> extracts in the usual fashion was followed by chromatography to give 10, as a foam, the yield

Table IV.	N-(Cvcloalkylmet)	$vl)-6\alpha.7\alpha$ (ox	ymethylene)-	7β-substituted-4,	$5\alpha$ -epoxymorphinans
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compd	R	R <sub>1</sub>	yield, %	mp, °C	recrystn solvent <sup>a</sup>	formula <sup>a</sup>
24Pc	C, H,	CH <sub>2</sub> -c-C <sub>3</sub> H <sub>5</sub>	60 <sup>b</sup>	144-146.5	EA	C <sub>30</sub> H <sub>35</sub> NO <sub>3</sub>
24Pd	CH <sub>2</sub> C <sub>4</sub> H	CH,-c-C,H,	93 <sup>d</sup>	152 - 157	E-W	$C_{31}H_{37}NO_3 \cdot C_4H_6O_6$
24 <b>Pf</b>	$(CH_{2})_{2}C_{6}H_{5}$	CH,-c-C,H,	99 <i>°</i>	110-115	E	$C_{32}H_{39}NO_3 \cdot C_4H_6O_6$
24 Bc	Ċ,H,	CH, -c-C <sub>4</sub> H,	99 <i>°</i>	137 - 146	E	$C_{31}H_{37}NO_3 \cdot C_4H_6O_6$
24Bd	CH <sub>2</sub> Č <sub>6</sub> H,	$CH_{2} - c - C_{4}H_{2}$	89 <sup>d</sup>	130 - 144	E	$C_{32}H_{39}NO_3 \cdot C_4H_6O_6$
24Bf	$(CH_{2})_{2}C_{6}H_{5}$	CH <sub>2</sub> -c-C <sub>4</sub> H <sub>2</sub>	99 <i>°</i>	128 - 134	E	$C_{33}H_{41}NO_3 \cdot C_4H_6O_6$
25Pc	C <sub>6</sub> H <sub>5</sub>	CH <sub>2</sub> -c-C <sub>3</sub> H <sub>5</sub>	$53^d$	188-190	$\mathbf{E}\mathbf{A}$	$C_{29}H_{33}NO_{3}$
25Pd	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	CH <sub>2</sub> -c-C <sub>3</sub> H <sub>5</sub>	$43^{d}$	foam		$C_{30}H_{35}NO_3$
25 Pf	$(CH_{2})_{2}C_{6}H_{5}$	CH,-c-C <sub>3</sub> H,	81 <i>°</i>	147–170 <i>°</i>	$\mathbf{E}\mathbf{A}$	$C_{31}H_{37}NO_3 \cdot C_4H_6O_6 \cdot 0.5EA$
25Bc	Ċ,H,	CH, -c-C <sub>4</sub> H,	$42^{d}$	200-202	$\mathbf{E}\mathbf{A}$	$C_{30}H_{35}NO_3$
25Bd	CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	CH,-c-C,H,	62 <sup>c</sup>	160-176	$\mathbf{E}\mathbf{A}$	$C_{31}H_{37}NO_{3}\cdot C_{4}H_{6}O_{6}$
25Bf	$(CH_2)_2C_6H_5$	$CH_2$ -c- $C_4H_7$	$82^{d}$	158-159	$\mathbf{E}$	$C_{32}H_{39}NO_{3}$

<sup>a</sup> See Table III for explanation. <sup>b</sup> Crystallized directly as the free base. <sup>c</sup> Crystallized directly as the salt. Yield of free base prior to salt formation. <sup>d</sup> Yield of free base after chromatography. <sup>e</sup> Foams.

being based on the NCH<sub>3</sub> compounds 7: 10c, 88%; 10d, 81%; 10f, 75%; 10g, 67%. These foams were used in the alkylation reactions described below.

 $7\beta$ -(Arylalkyl)-17-(cycloalkylmethyl)-4,5 $\alpha$ -epoxy-7 $\alpha$ -(hydroxymethyl)-3-methoxymorphinan-6 $\beta$ -ols (11P,B). A solution of 10 in DMF (1 g in 20 mL) containing NaHCO<sub>3</sub> (2.5 equiv) and cycloalkylmethyl bromide (1.2 equiv) was heated in an oil bath at 100 °C while under argon until the reaction was complete as indicated by TLC (3-20 h). The mixture was cooled and filtered to remove insolubles. The filtrate was evaporated with an oil pump, and the residue was dissolved in H<sub>2</sub>O. This mixture was adjusted to pH 10-11 with NH<sub>4</sub>OH and extracted with three portions of toluene. The organic phase was processed in the usual manner, the residue was chromatographed, and the product was crystallized as the free base or HCl salt as indicated in Table III.

 $7\beta$ -(Arylalkyl)-17-(cycloalkylmethyl)-4,5 $\alpha$ -epoxy-3hydroxy-7 $\alpha$ -(hydroxymethyl)morphinan-6 $\beta$ -ols (12P,B). A suspension of the 3-methoxy compound 11 in 48% HBr (1 g in 10 mL) was placed in a preheated, 140 °C oil bath, and the mixture was refluxed for 10-20 min. The cooled solution was diluted with H<sub>2</sub>O and made basic with NH<sub>4</sub>OH. This was extracted with EtOAc or CHCl<sub>3</sub>, the organic extracts were processed in the usual fashion, and the residue was chromatographed. Pure 12 was crystallized as the free base or HCl salt. Further details are given in Table III.

 $7\beta$ -(Arylalkyl)-4, $5\alpha$ -epoxy-3-methoxy-17-methyl- $6\alpha$ , $7\alpha$ -(oxymethylene)morphinans (18). The intermediates 17 were prepared from 16 by method A or method B described above for 5, followed by processing of the reaction mixture as detailed. Compound 17c was prepared by method A. Upon completion of the reaction with the phosphonium ylide, the mixture was poured into  $H_2O$ . The pH of this solution was adjusted to ca. 2, and the aqueous solution was extracted several times with toluene to remove  $(C_6H_5)_3PO$ . The aqueous phase was made basic with 50% NaOH, and 17c was extracted into CHCl<sub>3</sub>. A mixture of the two isomers of 17c was obtained in nearly quantitative yield. Hydrogenation of 17c was carried out in aqueous EtOH at 50 psi, with several changes of catalyst (10% Pd/C) until the reduction was complete as indicated by TLC. Workup gave 18c, as a foam, in 74% yield based on 16. This was converted to the d-tartrate salt, which was crystallized and recrystallized from EtOH to give pure 18c·d-tartrate, mp 192-197 °C. Anal. (C<sub>27</sub>H<sub>31</sub>NO<sub>3</sub>·C<sub>4</sub>H<sub>6</sub>O<sub>6</sub>) C, H, N. Intermediate 17d was prepared by method B, except that THF was used as the solvent for the addition of 16. Workup by selective extraction, followed by chromatography, gave a 44% yield of 17d. Hydrogenation for several days gave 18d as a glass in 68% yield from 17d. The d-tartrate salt, mp 191-192 °C, was obtained as crystals from MeOH-EtOAc. Anal. (C28H33NO3.  $C_4H_6O_6)$  C, H, N. The preparation of 18f, via 17f, has been reported.<sup>10</sup> Compound 17g was prepared by method A and obtained as a glass in 36% yield after chromatography. Hydrogenation gave 18g as a glass in 66% yield. Conversion to the d-tartrate salt, followed by crystallization from EtOH, gave an analytical sample of 18g.d-tartrate, mp 130-136 °C. Anal.  $(C_{30}H_{37}NO_{3}C_{4}H_{6}O_{6})$  C, H, N.

 $4,5\alpha$ -Epoxy- $7\beta$ - $(\alpha$ -hydroxybenzyl)-3-methoxy-17-methyl- $6\alpha,7\alpha$ -(oxymethylene)morphinan (19). A solution of 16 (1.72) g, 5.0 mmol) in toluene (100 mL) under argon was treated dropwise with phenyllithium (10 mmol). After the solution was stirred for 30 min, ice and H<sub>2</sub>O were added, and the organic phase was processed in the usual fashion to give 2.08 g (99%) of 19 as a foam. Chromatography allowed resolution of two isomers. The faster migrating isomer crystallized from EtOAc-Et<sub>2</sub>O to give 19, mp 222–223 °C. Anal. ( $C_{26}H_{29}NO_4$ ) C, H, N. The slower migrating isomer, 19, was crystallized from EtOAc to give needles, mp 237–240 °C. Anal. C, H, N.

 $7\beta$ -(Arylalkyl)-4, $5\alpha$ -epoxy-3-hydroxy-17-methyl- $6\alpha$ , $7\alpha$ -(oxymethylene)morphinans (21). A mixture of 18 and 48% HBr (1 g in 10-20 mL) was refluxed in a preheated oil bath for 15 min. The cooled solution was diluted with  $H_2O$  and made basic with NH<sub>4</sub>OH, and the  $6\alpha$ -hydroxy- $7\beta$ -bromomethyl intermediate 20 was extracted with CHCl<sub>3</sub> or EtOAc. After processing of the organic phase and evaporation, the residue was dissolved in dioxane (30 mL), and 1 N NaOH (10 mL) was added. The mixture was heated at 60-70 °C for 90 min and then evaporated to dryness. The residue was dissolved in  $H_2O$ , and the pH was adjusted to ca. 8 with HOAc. Extraction with CHCl<sub>3</sub> was followed by chromatography. Compound 21c was obtained as a foam in 40% yield. The d-hemitartrate, mp 200-208 °C, crystallized as the hemisolvate from EtOH. Anal.  $(C_{26}H_{29}NO_3 \cdot 0.5C_4H_6O_6 \cdot 0.5C_2H_6O) C_{28}$ H, N. In the same manner, 21d was obtained in 64% yield, and the d-tartrate salt, mp 234-236 °C, crystallized from aqueous EtOH to give hydrated material. Anal. (C27H31NO3 C4H6O6. 0.25H<sub>2</sub>O) C, H, N. Compound 21f was previously described. The free base 21g was obtained in 70% yield and converted to the d-tartrate salt. Several recrystallizations from EtOH gave the hydroscopic salt, mp 144-153 °C. Anal.  $(C_{29}H_{35}NO_3 \cdot C_4H_6O_6 \cdot C_{29}H_{35}NO_3 \cdot C_{4}H_6O_6 \cdot C_{29}H_{35}NO_3 \cdot C_{2}H_{35}NO_3 \cdot C_{2}H_{35}$ 0.25H<sub>2</sub>O) C, H, N.

 $7\beta$ -(Arylalkyl)-4,5 $\alpha$ -epoxy-3-methoxy-6 $\alpha$ ,7 $\alpha$ -(oxymethylene)morphinans (23). The N-cyano compounds were prepared as described above for 9. Compound 18c (16.6 g, 39.7 mmol) gave 16.3 g of 22c as a foam. This foam was dissolved in dioxane (500 mL), and KOH (465 mmol) in H<sub>2</sub>O (100 mL) was added. The mixture was refluxed for 1 week, cooled, and evaporated to a small volume. The residue was partitioned between  $H_2O$  and  $CHCl_3$ . Processing of the organic phase gave 16.0 g of a foam, which was chromatographed to give 12.2 g (76%) of 23cas a foam. Crystalline 23c, mp 101-102 °C, was obtained upon trituration with EtOH. Hydrolysis of 22d for 5 days, followed by workup, gave 23d in nearly quantitative yield as a glass, which was not further purified but used directly in alkylation reactions. Chromatography gave a 91% yield of 22f, which was hydrolyzed to 23f and purified by chromatography to give material suitable for further reactions.

 $7\beta$ -(Arylalkyl)-17-(cycloalkylmethyl)-4, $5\alpha$ -epoxy-3-methoxy- $6\alpha$ , $7\alpha$ -(oxymethylene)morphinans (24P,B). These compounds were prepared as described above for 11. Details are given in Table IV.

 $7\beta$ -(Arylalkyl)-17-(cycloalkylmethyl)-4,5 $\alpha$ -epoxy-3hydroxy-6 $\alpha$ ,7 $\alpha$ -(oxymethylene)morphinans (25P,B). These compounds were prepared as described above for 21. Details are given in Table IV.

 $7\beta$ -(Arylalkyl)-17-(cycloalkylmethyl)-4,5 $\alpha$ -epoxy-3-methoxy-7 $\alpha$ -methylmorphinan- $6\alpha$ -ols (26P,B). A suspension of AlCl<sub>3</sub>

(1.95 g, 14.6 mmol) in Et<sub>2</sub>O (100 mL) was cooled in an ice bath. and  $LiAlH_4$  (1.66 g, 43.9 mmol) was added. After the mixture was stirred for 30 min, a solution of 24f (7.1 g, 14.6 mmol) in Et<sub>2</sub>O (250 mL) was added, and the mixture was refluxed for 18 h. The cooled mixture was treated sequentially dropwise with  $H_2O$  (1.7 mL), 15% NaOH (1.7 mL), and H<sub>2</sub>O (5.1 mL) and then filtered through Celite. The filtrate was diluted with EtOAc and washed with  $H_2O$ . Evaporation gave 5.8 g of a glass, which was chromatographed to give 4.43 g (62%) of 26Pf as a glass. A portion of this material was converted to the HCl salt, which was obtained as white crystals, mp 118-122 °C, from EtOAc. Several recrystallizations from EtOAc gave analytically pure material with an indefinite melting point. Anal. (C<sub>32</sub>H<sub>41</sub>NO<sub>3</sub>·HCl) C, H, N. In a similar manner, 26Bc was obtained as a foam in 84% yield. Crystals of the HCl salt, mp 169-171 °C, were obtained from EtOAc. Anal. (C31H39NO3 HCl) C, H, N. Compound 26Bf was obtained in 65% yield and crystallized as the HCl salt: mp sinters at 116 °C, melts at 123-130 °C. Anal. (C33H43NO3 HCl) C, H, N.

 $7\beta$ -(Arylalkyl)-17-(cycloalkylmethyl)-4,5 $\alpha$ -epoxy-3-methoxy-7 $\alpha$ -methylmorphinan-6-ones (27P.B). To a solution of Me<sub>2</sub>SO (1.0 mL, 14.3 mmol) in CH<sub>2</sub>Cl<sub>2</sub> under argon at -60 °C was added slowly, dropwise, trifluoroacetic anhydride (1.5 mL, 10.7 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (7 mL). After 10 min, a solution of 26Pf (3.48 g, 7.1 mmol) in  $CH_2Cl_2$  (50 mL) was added slowly. The mixture was kept at -60 °C for 90 min, then TEA (3 mL) was added, and the mixture was warmed to room temperature. After the mixture was washed with  $H_2O$ , evaporation of the organic phase gave 3.47 g of 27Pf as a foam. The HCl salt, mp 105-110 °C, was recrystallized several times from EtOAc and best analyzed as containing 0.33 mol of EtOAc. Anal. (C32H39NO3 HCl-0.33EtOAc) C, H, N. The free base of **27Bc** was obtained as crystals, mp 150-151 °C, in 77% yield after chromatography and crystallization from EtOAc-Et<sub>2</sub>O. Anal. (C<sub>31</sub>H<sub>37</sub>NO<sub>3</sub>) C, H, N. Compound 27Bf was obtained in quantitative yield as crystals, mp 149-152 °C. Recrystallization from EtOAc gave a 96% yield of pure 27Bf, mp 150-151.5 °C. Anal. (C<sub>33</sub>H<sub>41</sub>NO<sub>3</sub>) C, H, N.

7β-(Arylalkyl)-17-(cycloalkylmethyl)-4,5α-epoxy-3hydroxy- $7\alpha$ -methylmorphinan-6-ones (28P,B). A mixture of 27 and 48% HBr was refluxed for 15 min and processed as described previously. Chromatography gave 28Pf in 63% yield as a foam. Crystals of 28Pf, mp 183.5–185 °C, were deposited from EtOAc. Anal. (C<sub>31</sub>H<sub>37</sub>NO<sub>3</sub>) C, H, N. Compound 28Bc was obtained in 42% yield after chromatography. The HCl salt, mp sinters above 200 °C, crystallized from EtOAc. Anal. (C<sub>30</sub>H<sub>35</sub>-

NO<sub>3</sub>·HCl) C, H, N. The foam obtained upon workup of 28Bf crystallized upon titration with EtOAc. These crystals, obtained in 53% yield, were recrystallized from EtOAc to give pure 28Bf, mp 186-188.5 °C. Anal. (C<sub>32</sub>H<sub>39</sub>NO<sub>3</sub>) C, H, N.

Registry No. 3, 85455-22-7; 5a, 85534-09-4; 5b, 85552-56-3; 5c, 85534-10-7; 5d, 85534-11-8; 5d·HCl, 85534-12-9; 5g, 85534-13-0; 5h, 85534-14-1; 5h-HCl, 85534-15-2; 6, 85534-16-3; 7a, 85534-17-4; 7a·HCl, 85534-18-5; 7b, 85534-19-6; 7b·HCl, 85534-20-9; 7c, 85534-21-0; 7d, 85534-23-2; 7d·HCl, 85534-22-1; 7f, 85455-23-8; 7g, 85534-25-4; 7g-HCl, 85534-24-3; 7h, 85534-26-5; 8b, 85534-27-6; 8c, 85534-28-7; 8d, 85534-29-8; 8f, 85455-24-9; 8g, 85434-31-7; 8g·HCl, 85534-30-1; 8h, 85534-33-4; 8h·HCl, 85534-32-3; 10c, 85534-34-5; 10d, 85534-35-6; 10f, 85534-36-7; 10g, 85534-37-8; 11Bc, 85552-58-5; 11Bc·HCl, 85552-57-4; 11Bd, 85534-45-8; 11Bf, 85534-47-0; 11Bf·HCl, 85534-46-9; 11Bg, 85534-49-2; 11Bg·HCl, 85534-48-1; 11Pc, 85534-39-0; 11Pc·HCl, 85534-38-9; 11Pd, 85534-40-3; 11Pf, 85534-42-5; 11Pf·HCl, 85534-41-4; 11Pg, 85534-44-7; 11Pg·HCl, 85534-43-6; 12Bc, 85534-56-1; 12Bd, 85534-58-3; 12Bd·HCl, 85534-57-2; 12Bf, 85534-60-7; 12Bf·HCl, 85534-59-4; 12Bg, 85534-62-9; 12Bg·HCl, 85534-61-8; 12Pc, 85534-50-5; 12Pd, 85534-52-7; 12Pd·HCl, 85534-51-6; 12Pf, 85534-53-8; 12Pg, 85534-55-0; 12Pg·HCl, 85534-54-9; 16, 85455-05-6; (E)-17c, 85534-63-0; (Z)-17c, 85534-64-1; 17d, 85534-65-2; 17f, 85534-66-3; 17g, 85534-67-4; 18c, 85534-68-5; 18c d-tartrate, 85534-69-6; 18d, 85534-70-9; 18d d-tartrate, 85534-71-0; 18f, 85455-07-8; 18g, 85534-72-1; 18g d-tartrate, 85534-73-2; 19 (isomer 1), 85534-74-3; 19 (isomer 2), 85534-75-4; 20c, 85534-76-5; 20d, 85534-77-6; 20g, 85552-59-6; 21c, 85534-78-7; 21c d-hemitartrate, 85534-79-8; 21d, 85534-80-1; 21d d-tartrate, 85534-81-2; 21f, 85455-08-9; 21g, 85534-82-3; 21g d-tartrate, 85534-83-4; 22c, 85534-84-5; 22d, 85534-85-6; 22f, 85534-86-7; 23c, 85534-87-8; 23d, 85534-88-9; 23f, 85534-89-0; 24Bc, 85534-90-3; 24Bd, 85534-91-4; 24Bf, 85534-92-5; 24Pc, 85534-93-6; 24Pd, 85534-94-7; 24Pf, 85534-95-8; 25Bc, 85534-96-9; 25Bd, 85534-97-0; 25Bf, 85534-98-1; 25Pc, 85534-99-2; 25Pd, 85535-00-8; 25Pf, 85535-01-9; 26Bc, 85535-02-0; 26Bc·HCl, 85535-03-1; 26Bf, 85535-04-2; 26Bf·HCl, 85535-05-3; 26Pf, 85535-06-4; 26Pf·HCl, 85535-07-5; 27Bc, 85535-08-6; 27Bf, 85535-09-7; 27Pf, 85535-10-0; 27Pf·HCl, 85535-11-1; 28Bc, 85535-12-2; 28Bc·HCl, 85535-13-3; 28Bf, 85535-14-4; 28Pf, 85535-15-5; (C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>PCH<sub>3</sub>Br, 1779-49-3; (C<sub>6</sub>- $H_{5}_{3}P(CH_{2})_{2}CH_{3}Br, 6228-47-3; (C_{6}H_{5})_{3}PCH_{2}C_{6}H_{5}Br, 1449-46-3; (C_{6}H_{5})_{3}P(CH_{2})_{2}C_{6}H_{5}Br, 53213-26-6; (C_{6}H_{5})_{3}P(CH_{2})_{4}C_{6}H_{5}Br,$ 37748-19-9; (C<sub>6</sub>H<sub>5</sub>)<sub>3</sub>P(CH<sub>2</sub>)<sub>5</sub>C<sub>6</sub>H<sub>5</sub>Br, 17483-25-9; phenyllithium, 591-51-5.

## 2-(6-Carboxyhexyl)cyclopentanone Hexylhydrazone. A Potent and Time-Dependent **Inhibitor of Platelet Aggregation**

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Two new azaprostanoids, a hydrazone (3) and hydrazide (4), have been prepared by the condensation of 2-(6carboxyhexyl)cyclopentanone with n-hexylhydrazine and caproic acid hydrazide. Preliminary results with the stable hydrazide 4 indicate that it inhibits arachidonic acid (AA) induced human platelet aggregation and that, unlike 13-azaprostanoic acid (1), its site of action is at the cyclooxygenase level. Results with the unstable hydrazone derivative 3 indicate it to be a potent and time-dependent inhibitor of AA-induced human platelet aggregation, with its site of action also at the cyclooxygenase level.

In a continued search for modulators of platelet prostaglandin action, we prepared the azaprostanoid derivatives 3 and 4. The rational for the synthesis of these compounds was based on our previous observations of antiplatelet activity for the simple trans-13-azaprostanoic acid (1).<sup>1</sup> This derivative appears to act as a direct thromboxane antagonist at the receptor level.<sup>2</sup> We



speculated that the nonbonding electrons of nitrogen in the 13-azaprostanoic acid might be mimicking the  $\pi$  system

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